

SMS 204: Integrative Marine Sciences, Physics & Chemistry

- Instructors: E. Boss & L. Karp-Boss
- Office hours: 3-4 on Mon. Aubert 458
- Significant digits, graphs.
- Class web site:
http://misclab.umeoce.maine.edu/boss/classes/SMS_204/Syllabus.htm

- Some important concepts from last week:
 - No-slip
 - Treatment of fluid as Continuum (1ml $\sim 3 \times 10^{22}$ water molecules)
 - Viscosity
 - Density of solids

This week:

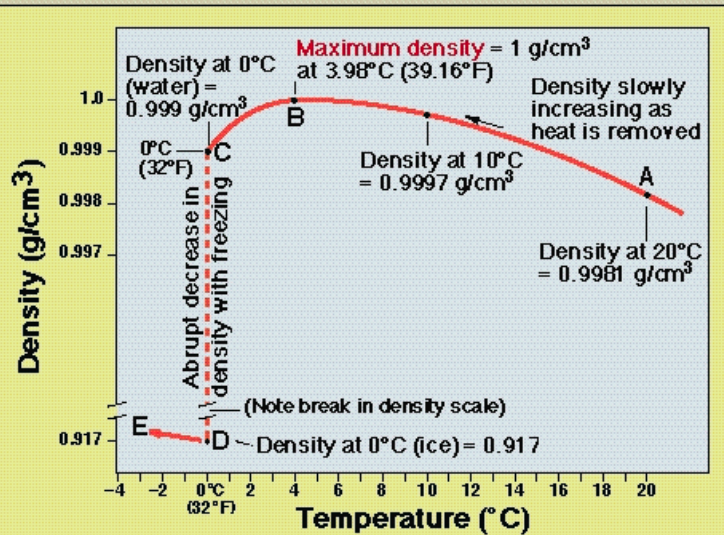
- Density of water
- Mass and volume fluxes
- Ideal gas
- Pressure

Density of Water

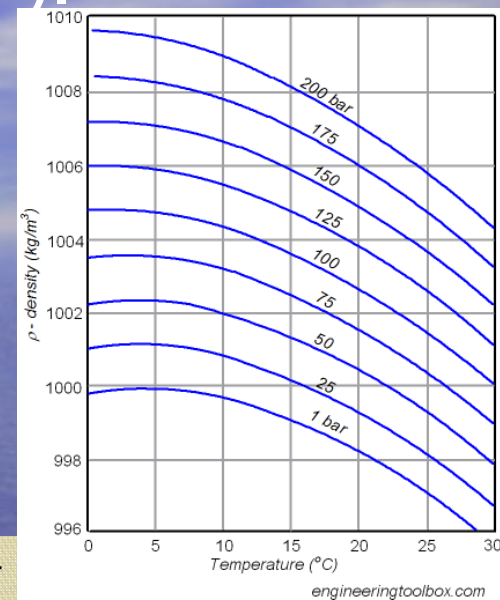
- Density = mass / volume
- Density of water depends on temperature
- Density of water depends on salinity
- Density of water depends on pressure
- Dense water sinks under less dense water

Equation of state of water: relates density to other variables (S, T & P).

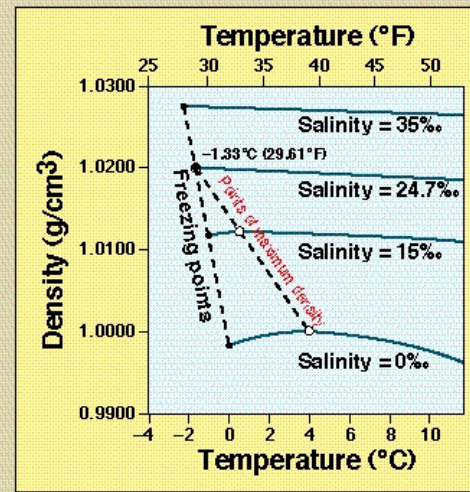
Density vs. Temperature for Pure Water



Density vs. T and P:



Salinity vs. Freezing Point of Water



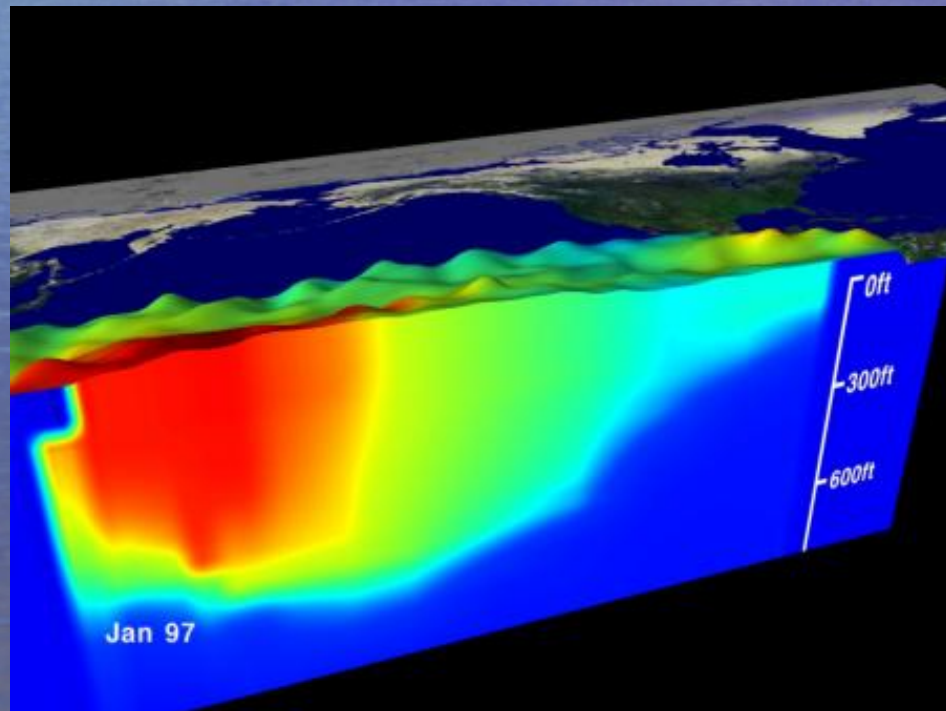
The dependence of freezing temperature and temperature of maximum density upon salinity. Pure water is densest at 3.98 degrees C (39.16 degrees F), and its freezing point is 0 degrees C (32 degrees F). Seawater with 15‰ salinity is densest at 0.73 degrees C (33.31 degrees F), and its freezing point is -0.80 degrees C (30.56 degrees F). The temperature of maximum density and freezing point coincide at -1.33 degrees C (29.61 degrees F) in seawater with a salinity of 24.7 ‰. At salinities greater than 24.7 ‰, the density of water always decreases as temperature increases.

From:
<http://geoserv.geology.wmich.edu/dave>

Density and water movement:

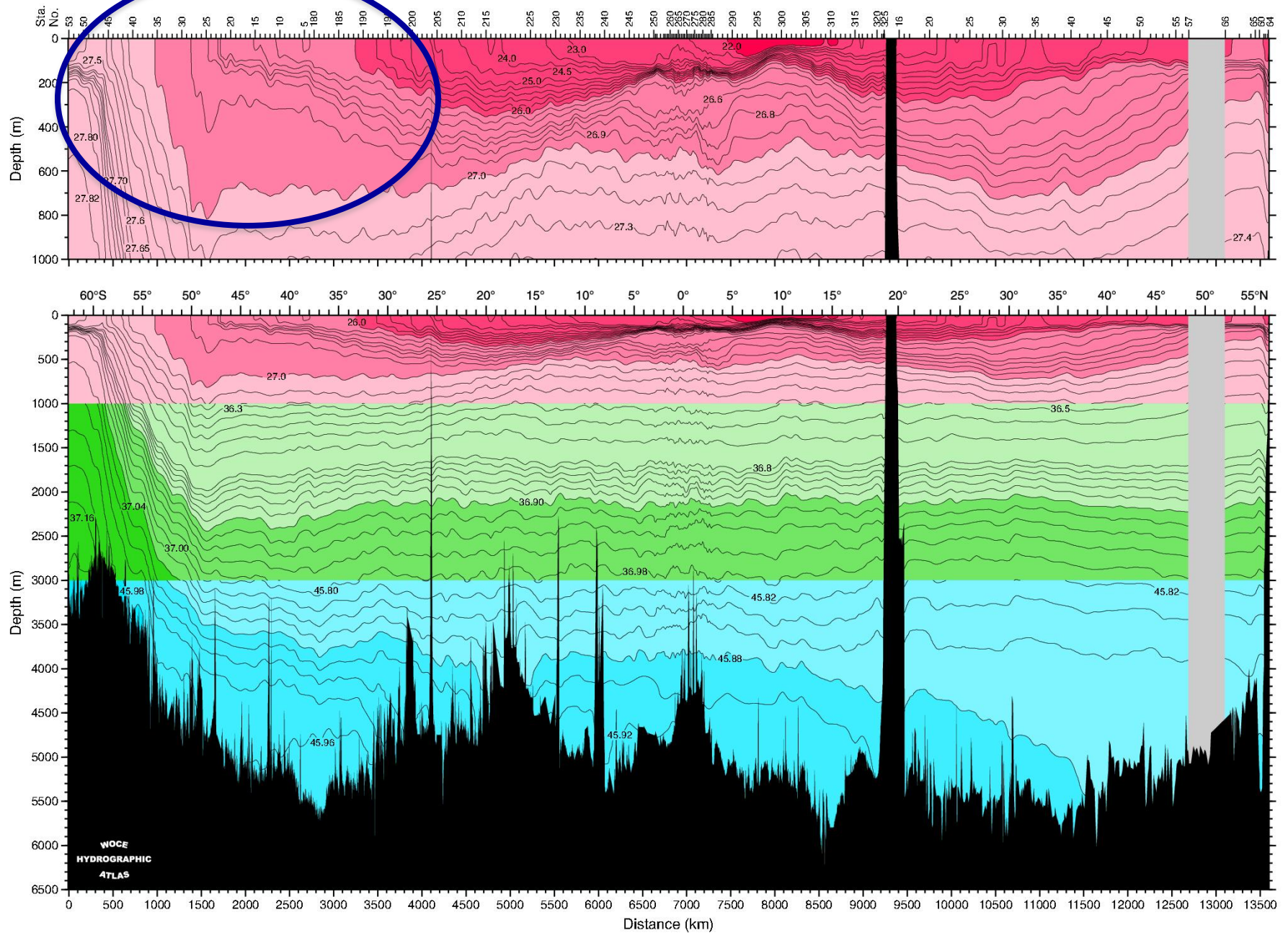
What happens when:

1. Dense water is above less dense water?
2. Dense water is next to less dense water? ← Demo
3. What about:



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sciences

Density stratification across the Pacific Ocean



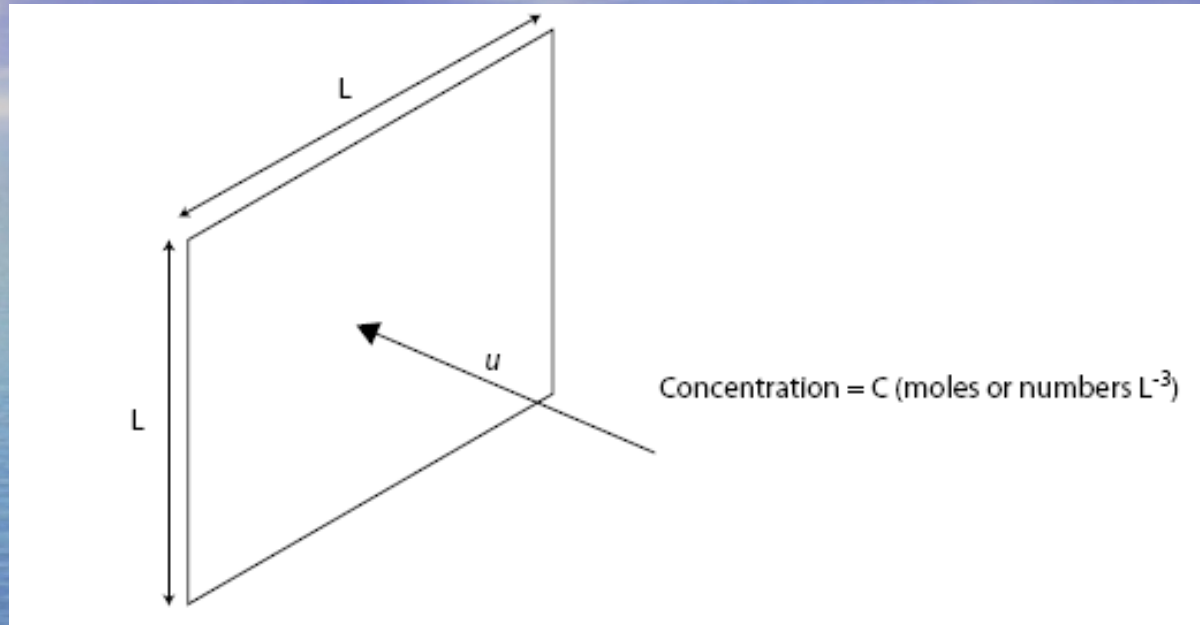
Flux: movement of fluids

- Volume of moving water
- Mass of moving water
- Momentum or heat of moving water
- Matter in moving water

Flux through a channel (or blood vessel)

- Fluid passing through a certain area at a given time.
- Volume flux: $\text{cross-section}(\perp) \times \text{velocity}$
- Mass flux: Volume flux \times density
- Material flux: Mass flux \times concentration

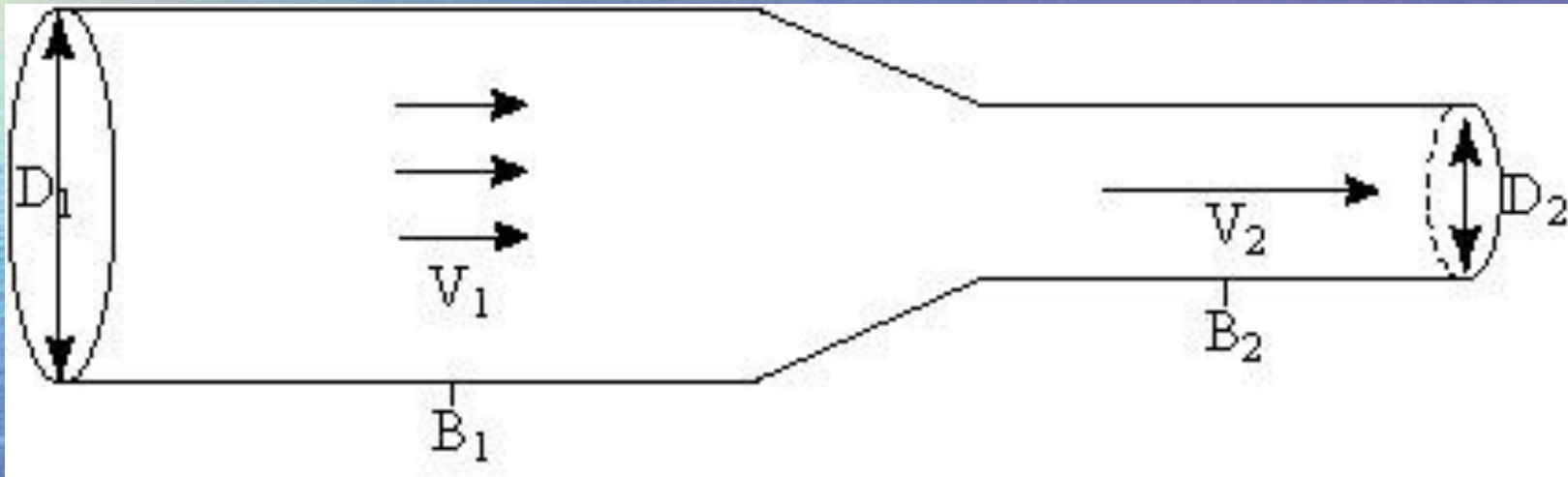
- Advective flux of matter (molecules, fish)



Advective flux = Area x Velocity x Concentration

$$\text{mol T}^{-1} = \text{L}^2 \times \text{L T}^{-1} \times \text{mol L}^{-3}$$

- Mass, volume and density.
 - In fluids we often cannot follow a coherent mass.
- Conservation of mass is described by mass continuity (incompressible flows):



- Volume flux: vA [m^3/sec], $A \perp v$
- Mass flux: ρvA [Kg/sec], $A \perp v$
- How do you get a hose to squirt further?

- Newton's laws of motion:
 - Without force a body will continue its motion
 - $d(\text{momentum})/dt = \text{Force}$
 - When a body 1 applies a force on body 2, an equal and opposite force is applied on body 1 by body 2.
 - In continuum mechanics, this formulation generalizes body 1 and its surrounding medium.
- Momentum = mv
- Momentum flux = mass flux $\times v =$
 $\rho v^2 A \text{ [Kg m s}^{-2}\text{]}, A \perp v$

Note: v -velocity, V -volume

- Pressure (P):

- Force per unit area
- Sharp vs. blunt objects
- Hydrostatic pressure: the weight (divided by A) of the fluid above-

$$P = Mg/A = \rho gh \text{ (for constant } \rho \text{)}$$

Pressure (depth) = density x depth x gravity + P_{air}

Problem solving in class: force on a diver's face

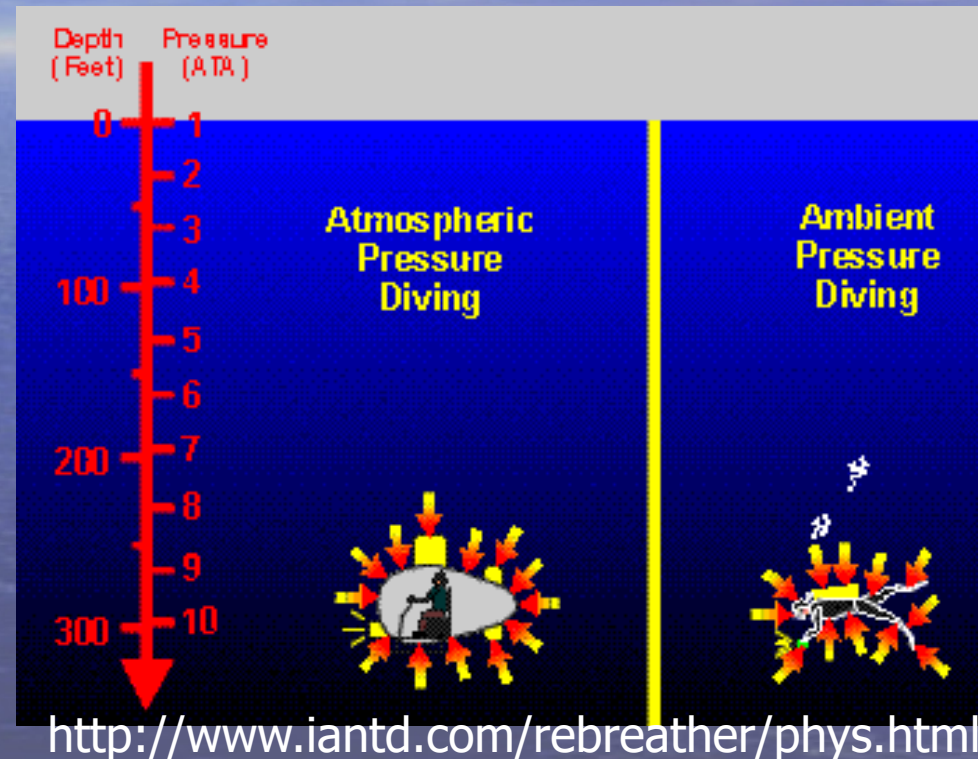
In class demonstrations

- Pressure vs. depth
 - Hydrostatic pressure: $p = \rho gh$ (for constant ρ)
 - In the ocean ρ changes by only a few percents.
- $h = p / \rho g$,
- Depth in oceanography is often denoted by pressure.

The pressure of 1 dbar = 10,000 Pa is similar to 1 m of water.

Diving and pressure

- Equalizing pressure in cavities.
- Why we can't snorkel deep? Why does SCUBA work?
- Gas solubility and pressure.
- How do some marine mammals prevent lungs from collapsing?



Fluid moves from regions of high to regions of low pressure:

$$F=ma \rightarrow dv/dt=F/m=F/(AL\rho)$$
$$\rightarrow dv/dt=-\rho^{-1}dp/dx$$

If fluid is not moving – pressure is equal.

If I have dense fluid at one side of a container and less dense at the other. How will pressure be distributed?

Equation of state of an ideal gases

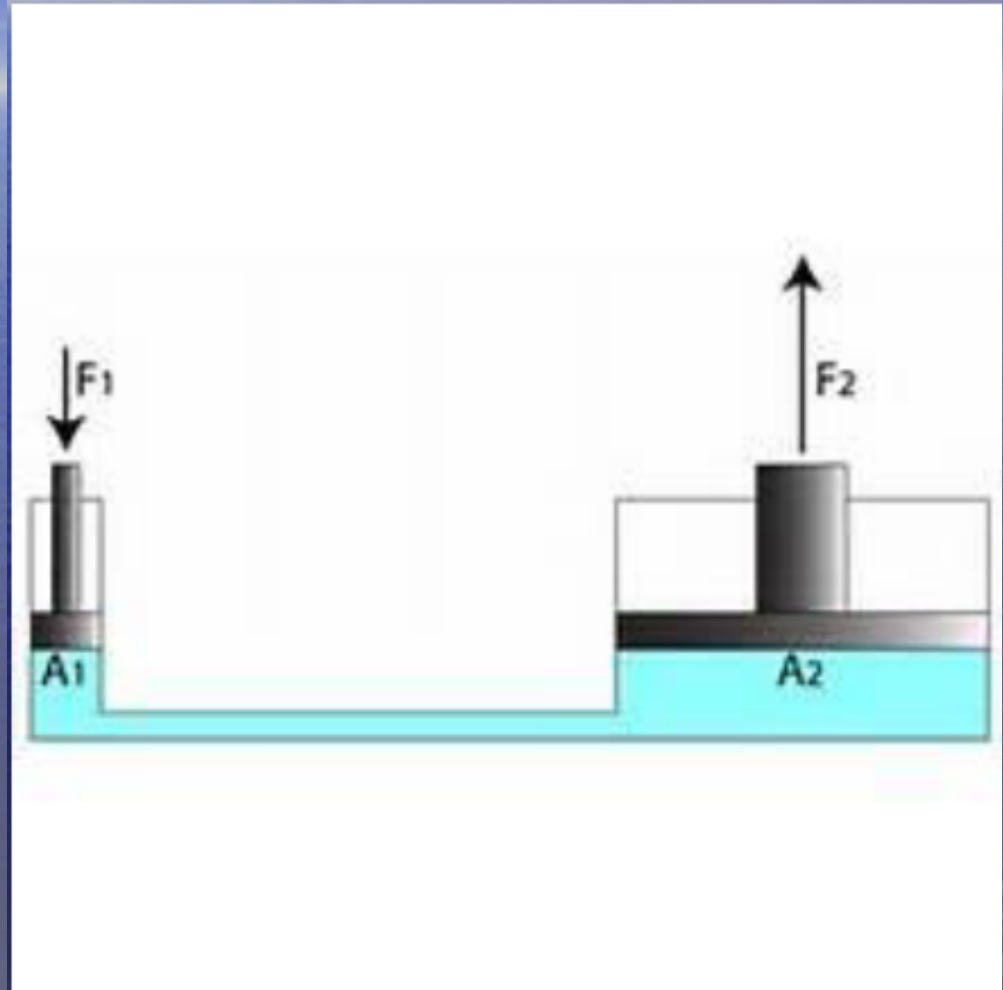
$$PV=nRT$$

- Temperature: kinetic energy of molecules.
- Pressure: momentum transfer (normal) to sides of container.
- Viscosity: momentum transfer between molecules.

In class demonstration (change of volume with pressure)

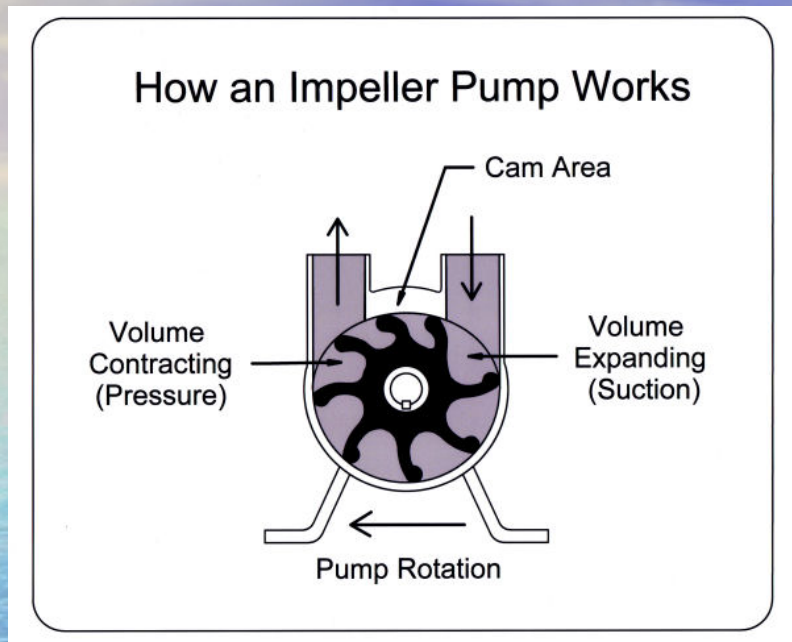
The magic of the hydraulic press:

- Pressure (non-hydrostatic) is equal within the fluid.
- $\rightarrow F_1/A_1 = F_2/A_2$
- $\rightarrow F_2 = F_1 A_1/A_2$
- $\rightarrow F_1 < F_2$
- We can use a small force to lift a heavy object if we apply it over a long distance (remember $\text{work} = \text{force} \times \text{distance}$).

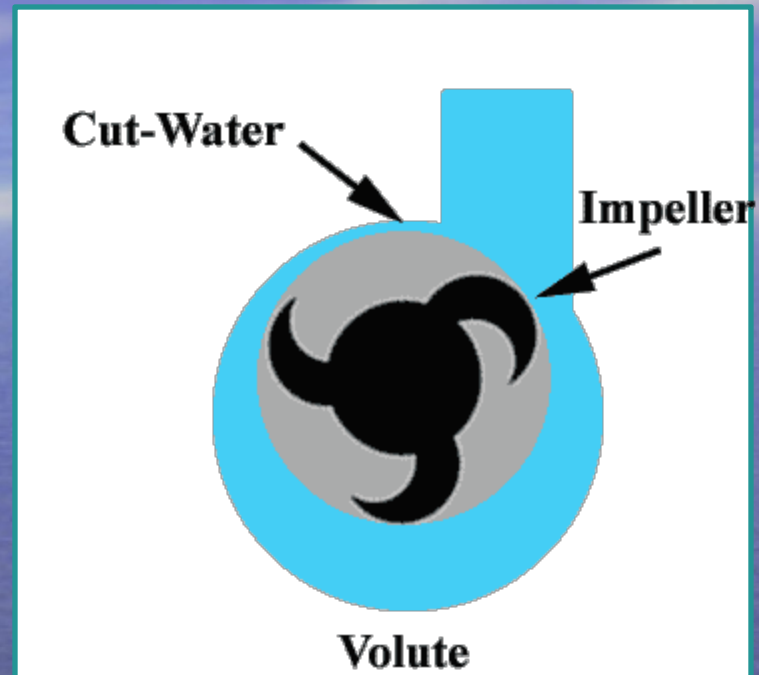


Two kinds of pumps:

- Positive displacement pump (decrease in volume raises pressure , *e.g.* a bicycle pump, diaphragm pump, peristaltic pump)
- Fluid dynamic pump (add thrust to the fluid through moving parts).
- Positive displacement pumps tend to be better at producing high pressures. Fluid dynamic pumps are better at producing large volumetric flow rates.



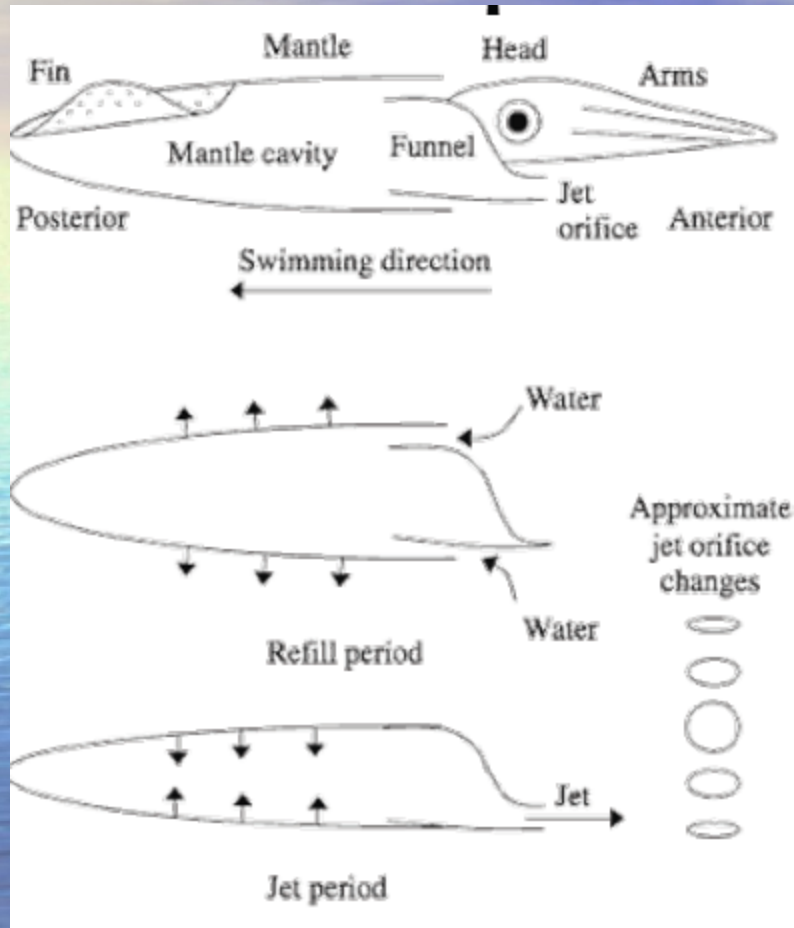
From: <http://captnpauley.typepad.com>



<http://www.perfusionkorea.org/ko/sect/img/ImpellerAnim.gif>

Which kind of pump are these?

Pumps in organisms:



Wikimedia

[Movie](#)

Summary

- Water organizes itself by density if there are no forces acting on it.
- Water flows from high to low pressure.
- Pumps are useful as tools to get food and get rid of waste products.
- If you care about diving (or organisms that dive) you should worry about pressure.