

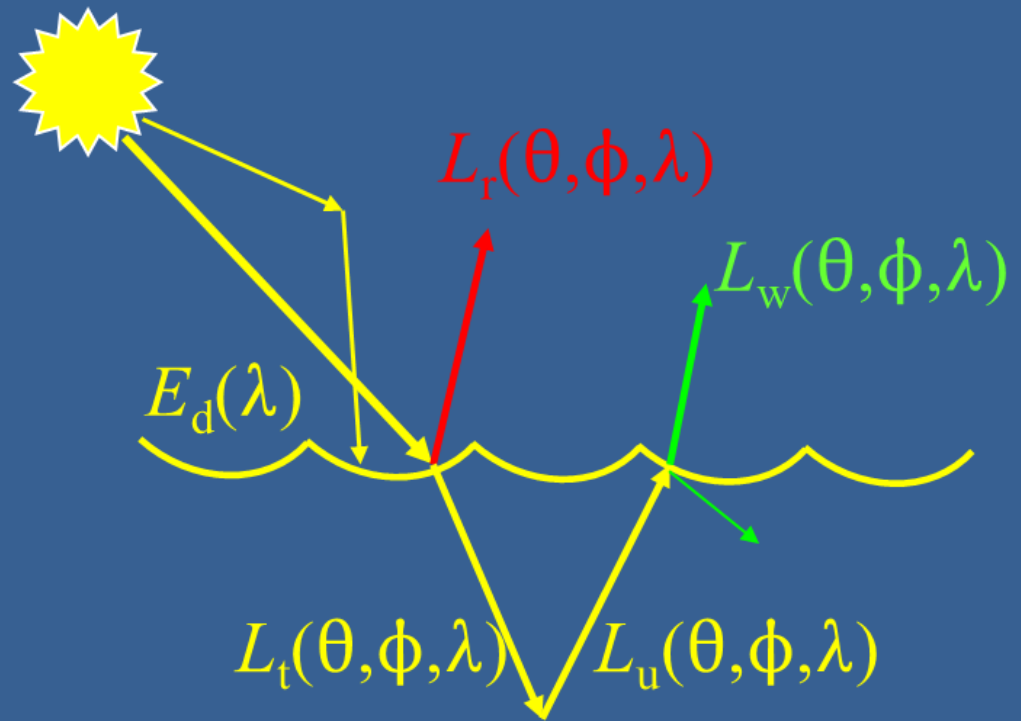
# Multiple Choice Exam on Chlorophyll Fluorescence

(allowing you to show off all that you have learned this week)

with answers

The purpose of this exercise is to show  
why you work with  $R_{rs}$  and not  $L_w$ ,  
and to get some experience analyzing  $R_{rs}$  spectra

Quick review of  $R_{rs}$ :  
The apparent optical  
property (AOP) used  
for remote sensing

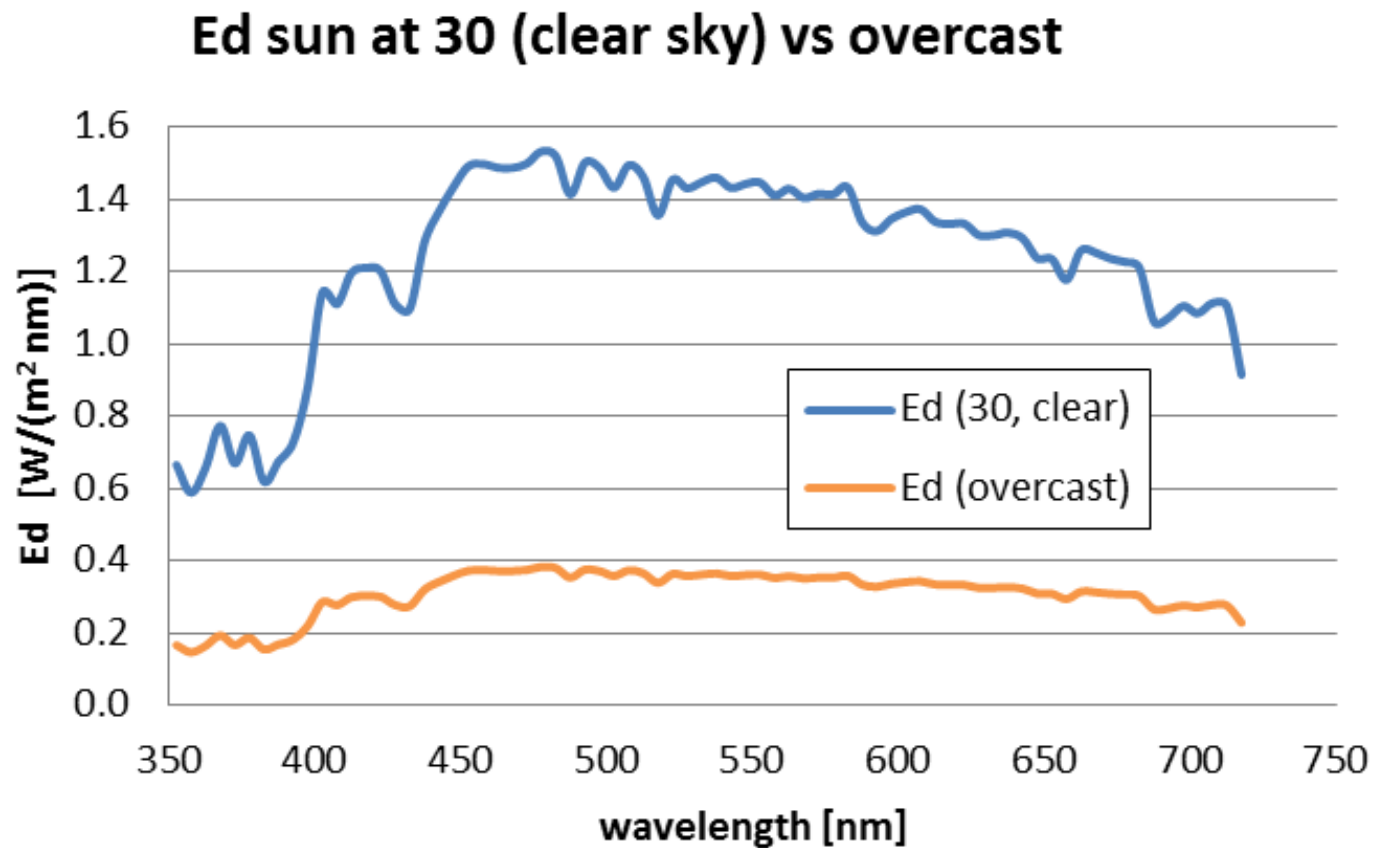


$$R_{rs}(\theta, \varphi, \lambda) =$$

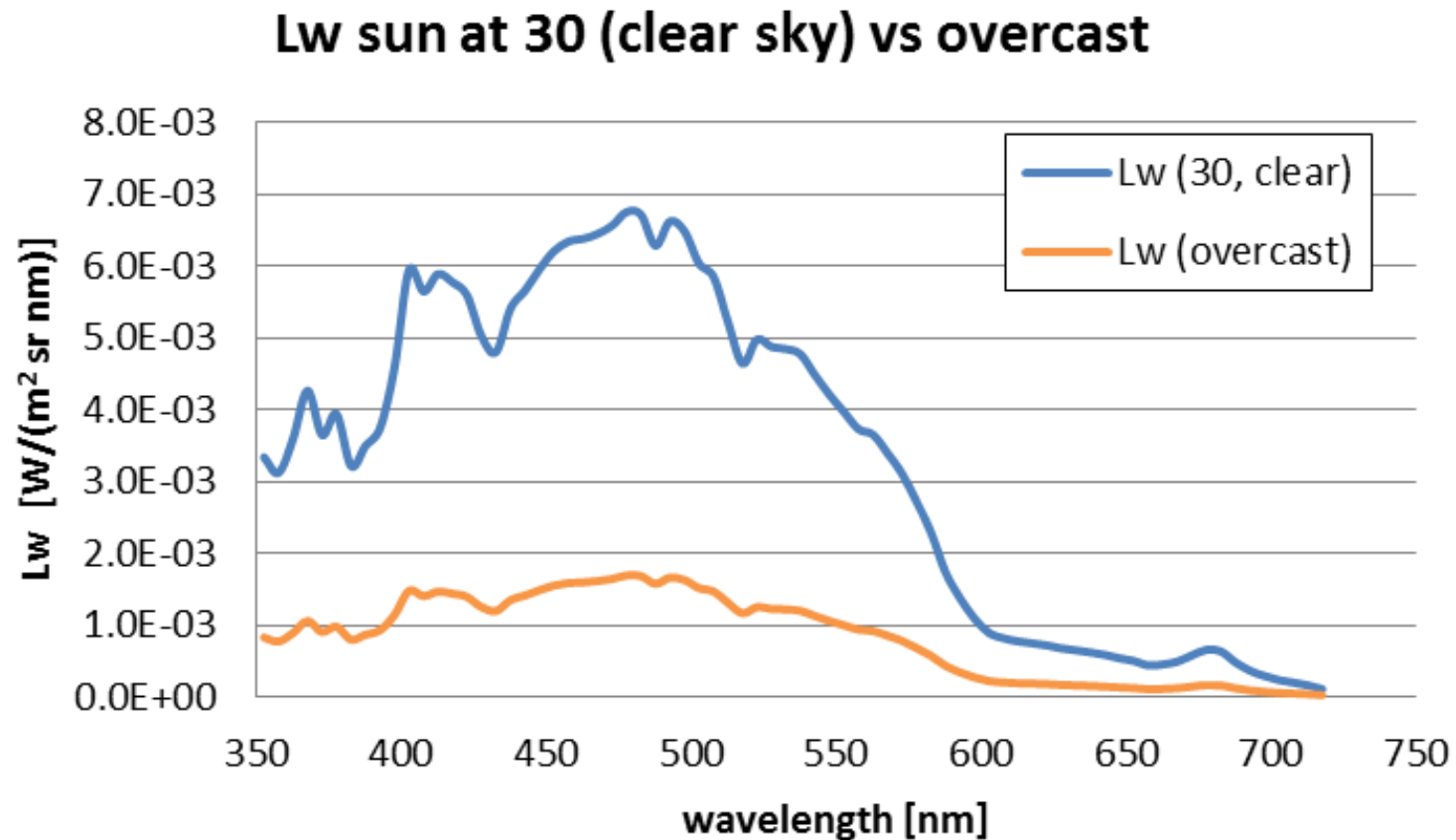
upwelling water-leaving radiance  
downwelling plane irradiance

$$R_{rs}(\text{in air}, \theta, \varphi, \lambda) \equiv \frac{L_w(\text{in air}, \theta, \varphi, \lambda)}{E_d(\text{in air}, \lambda)} \quad [\text{sr}^{-1}]$$

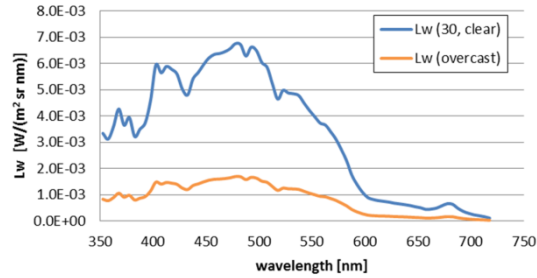
HydroLight run with sun at a 30 deg zenith angle in a clear sky vs an overcast sky (5 nm resolution,  $U = 5$  m/s, RTE solved to 30 m, etc.)



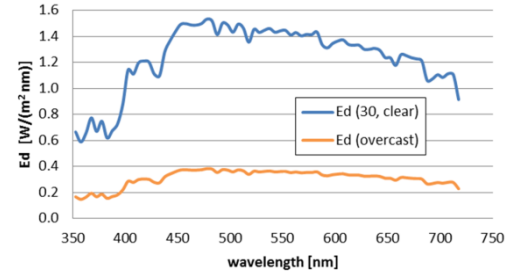
# HydroLight run with sun at a 30 deg zenith angle in a clear sky vs an overcast sky



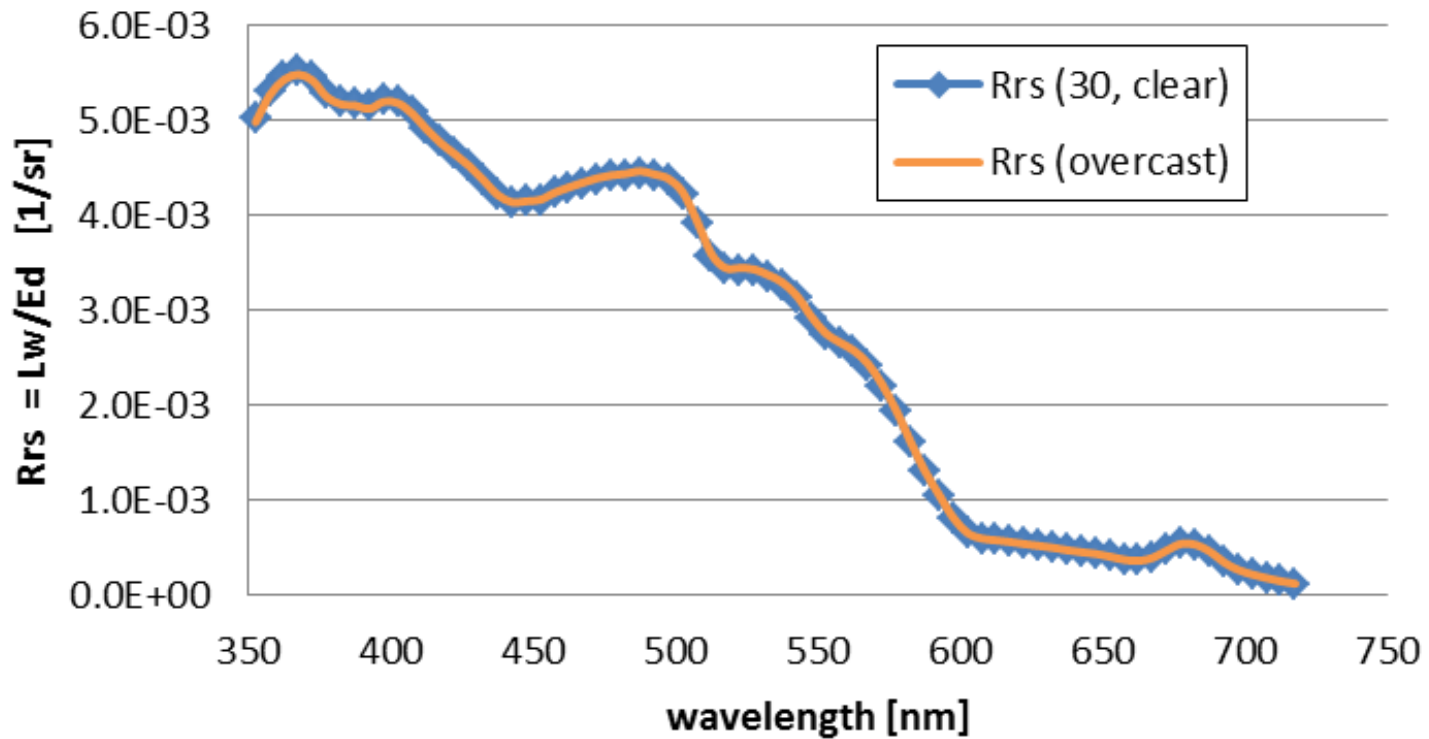
Lw sun at 30 (clear sky) vs overcast



Ed sun at 30 (clear sky) vs overcast



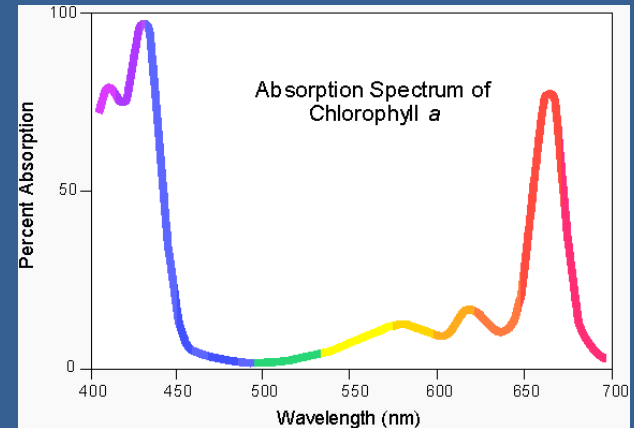
$R_{rs}$  sun at 30 (clear sky) vs overcast



Remember that Chl fluorescence is proportional to

- How much chlorophyll there is to absorb light
- How much light is available to be absorbed
- How efficiently the chlorophyll re-emits photons (the quantum efficiency)

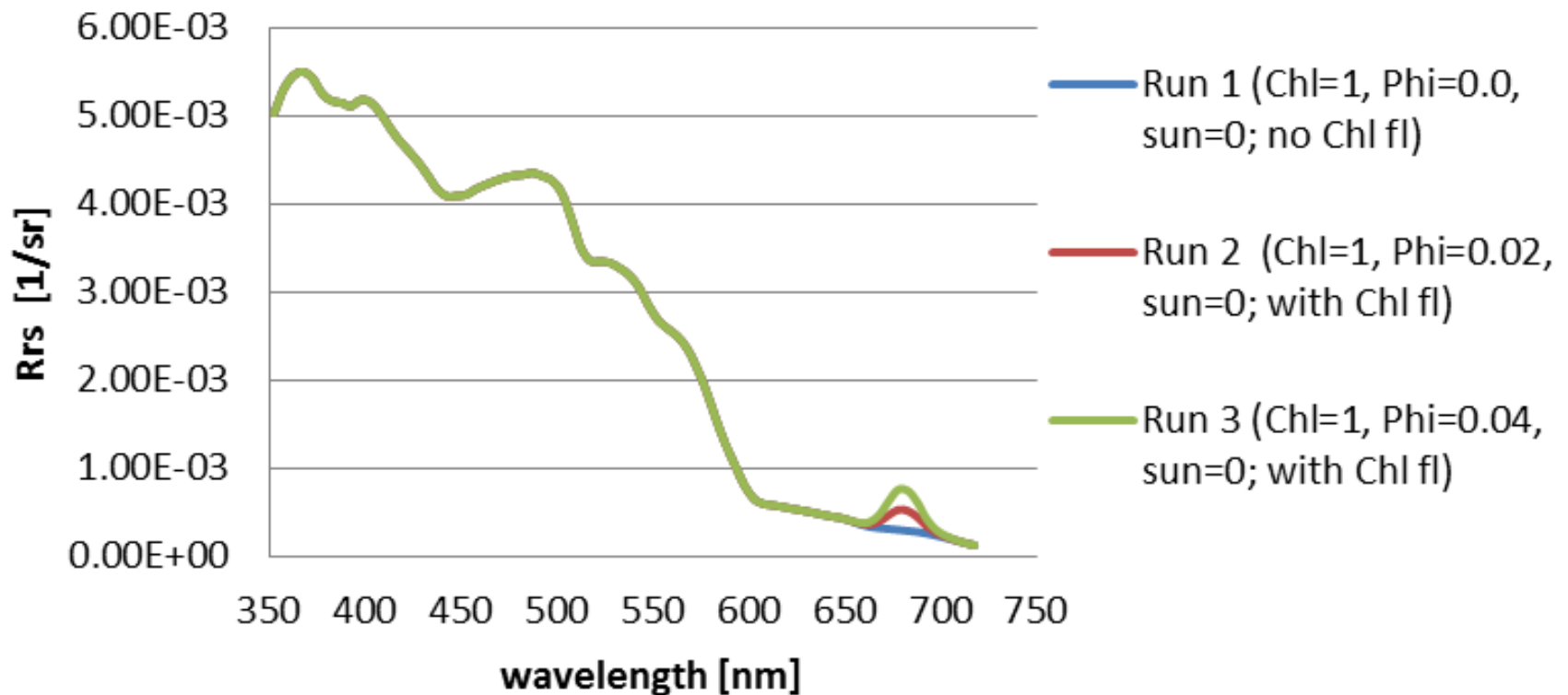
Now you explain the following  $R_{rs}$  spectra. Your choices are



From MJP

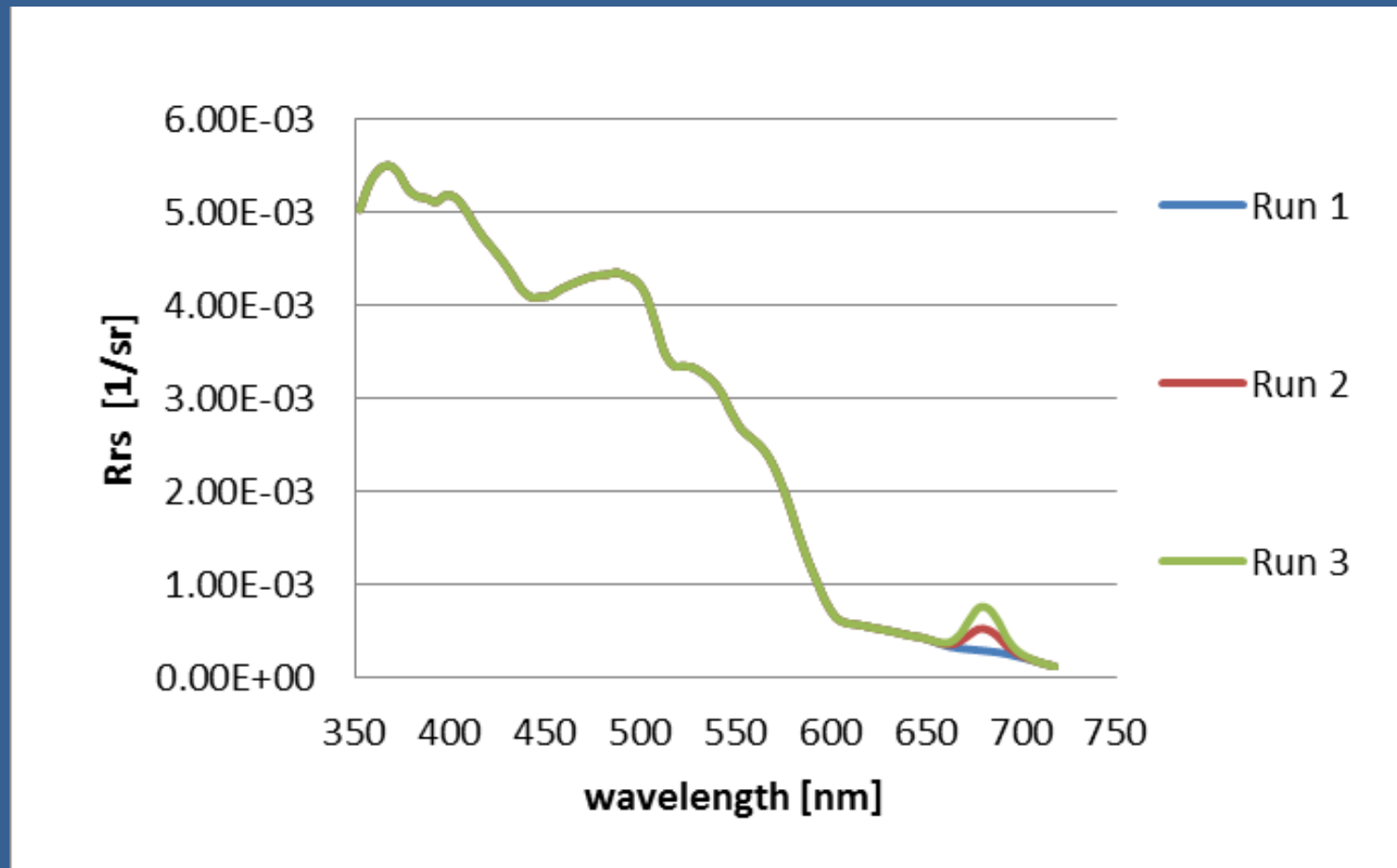
- Chl =  $1 \text{ mg m}^{-3}$  vs  $5 \text{ mg m}^{-3}$
- Quantum Efficiency  $\Phi_{\text{chl}} = 0$  (no chl fl) vs 0.02 vs 0.04
- Sun zenith angle = 0 vs 60 deg

# What is the Chl value : 1 or 5?



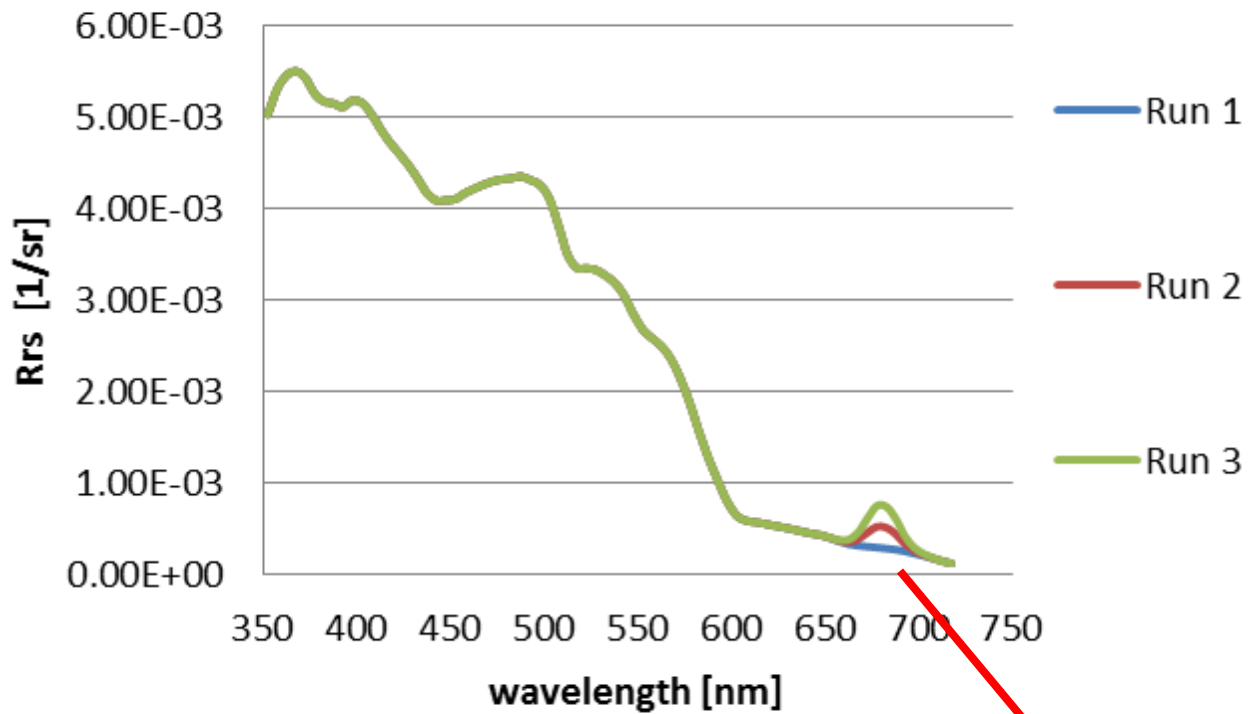
This is a very blue spectrum, so low Chl. Chl = 1 rather than 5, which would be green water

# What is the sun zenith angle: 0 or 60 deg?



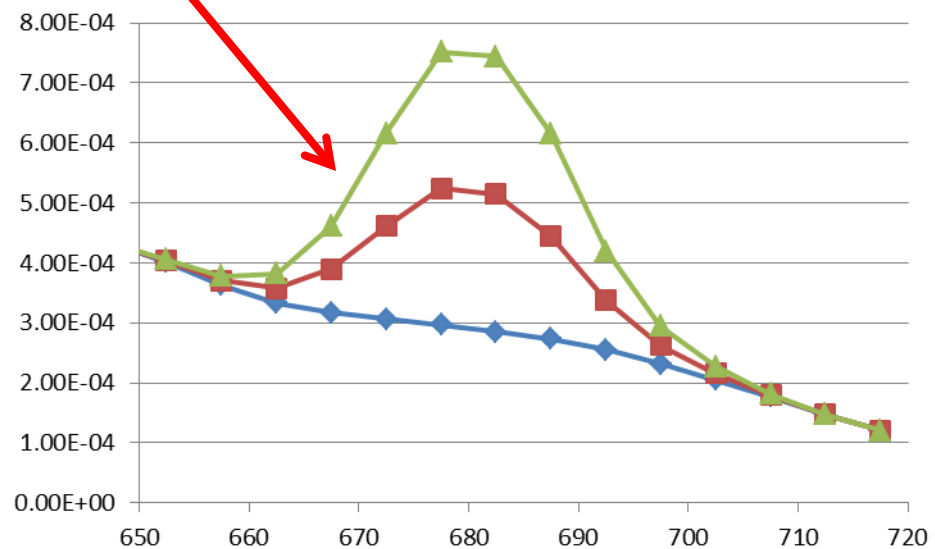
You can't say anything about the sun angle from  $R_{rs}$ ; that's the whole reason for using an AOP like  $R_{rs}$ , which normalizes out the sky effects



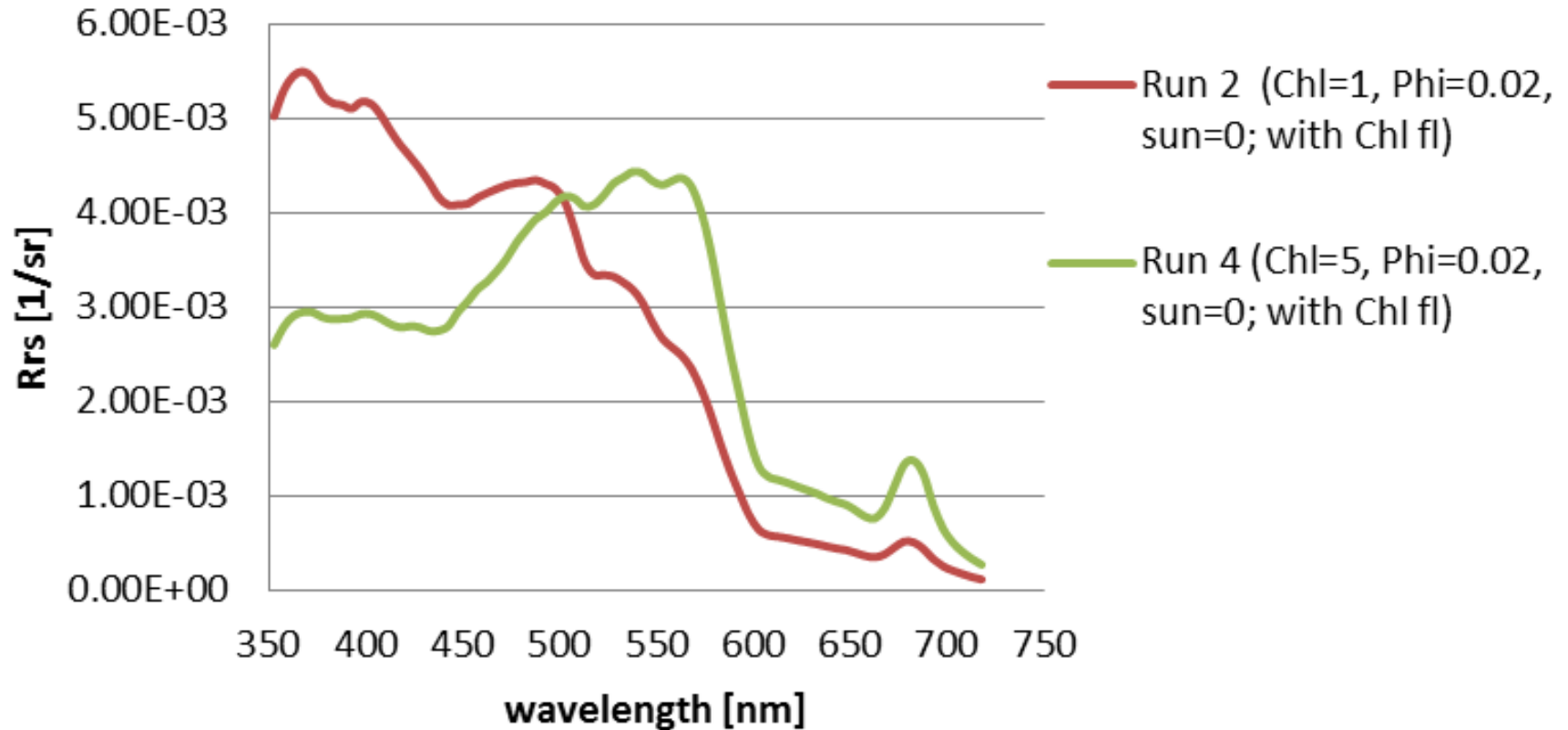


What causes these differences?

Different Phi values, since only the chl fluorescence band is different

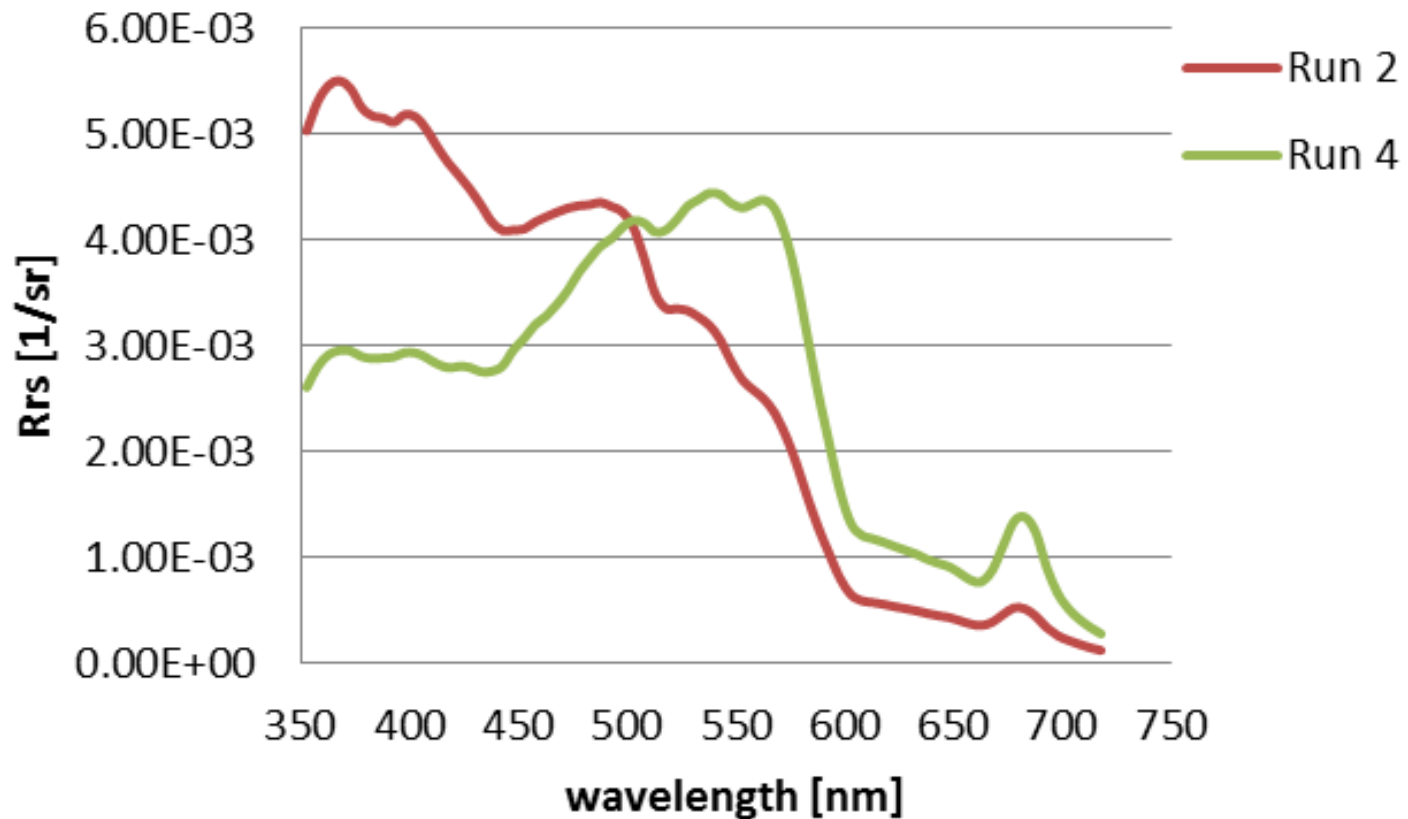


# What causes these differences?



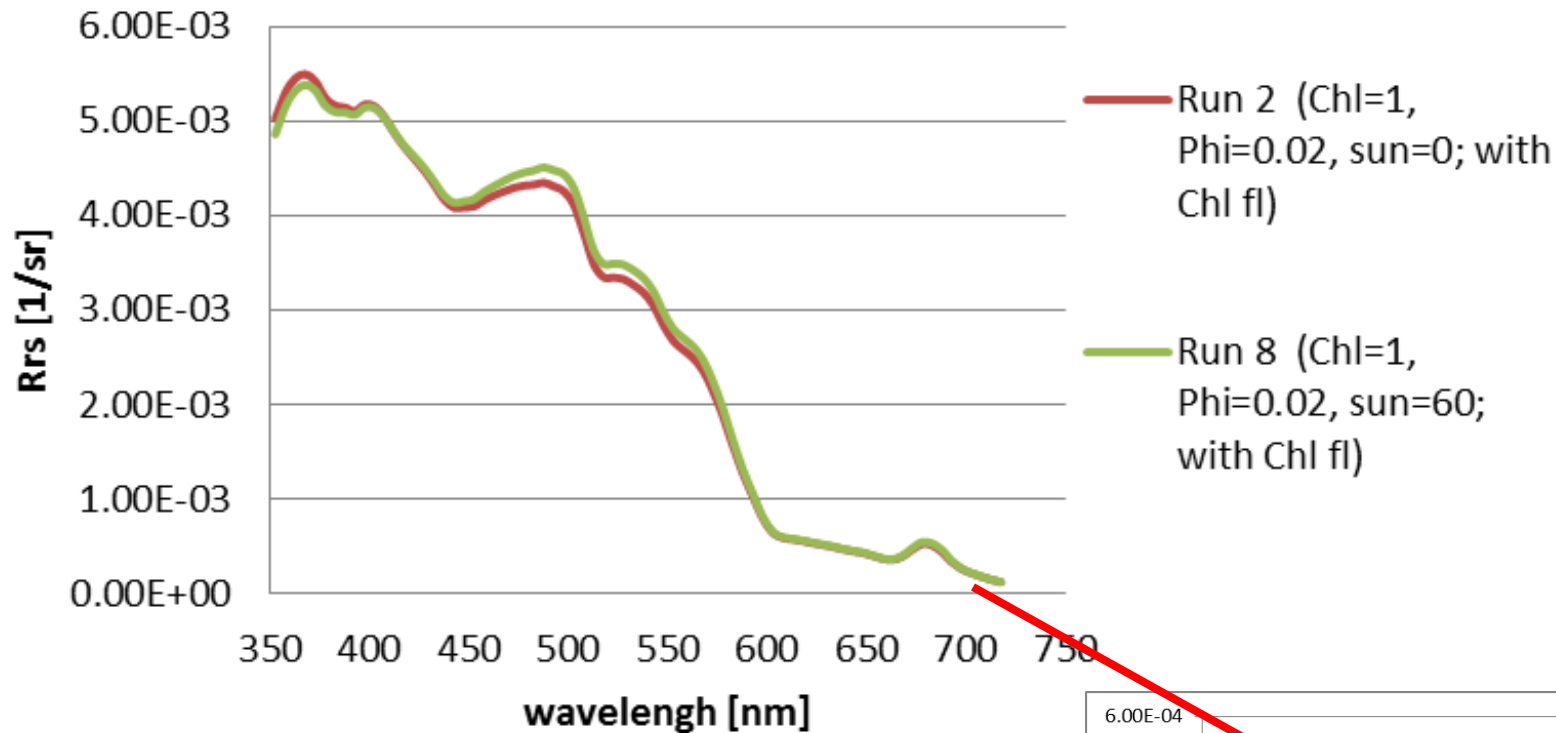
These are low (blue water) vs high (green water) Chl values

# What is the value of $\Phi_{\text{chl}}$ ?

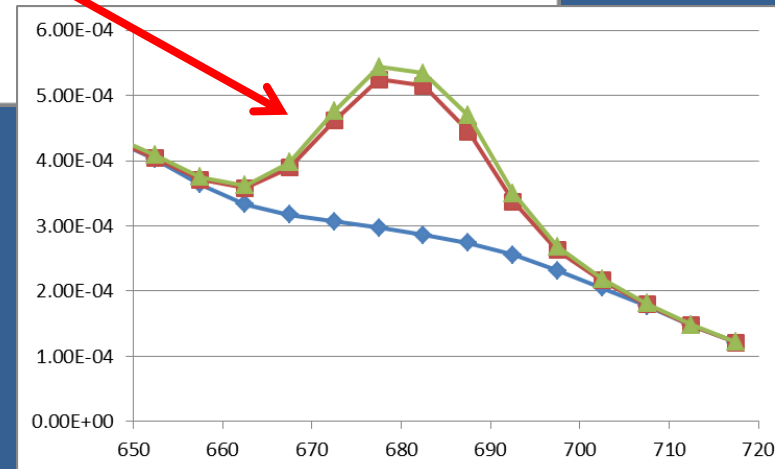


You can't really say much about the Phi value without additional information

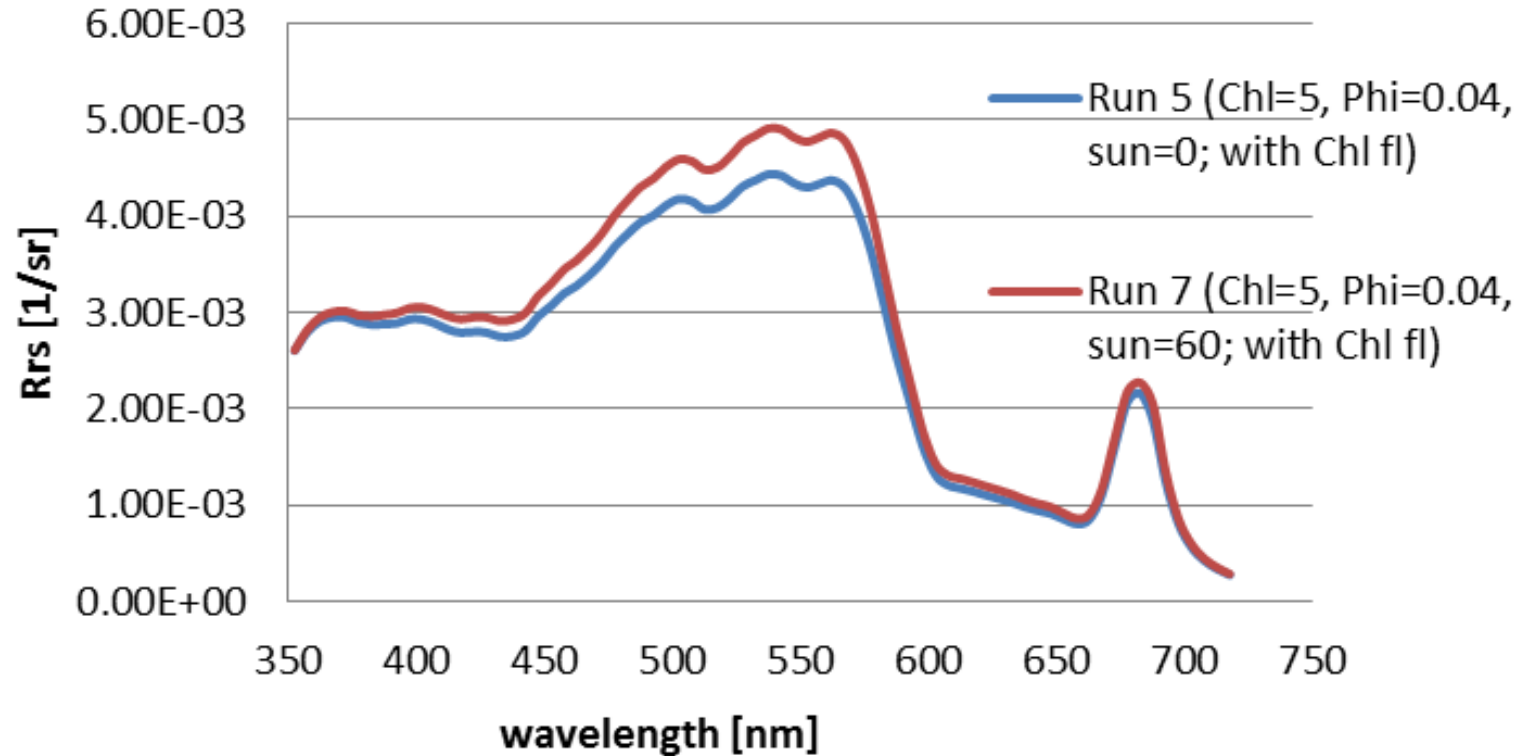
# What causes these differences?



These are sun angle differences, because the differences cover the whole spectrum, not just the Chl fl band. The two sun angles are sampling different parts of the VSF, with complicated interactions of absorption and scattering.

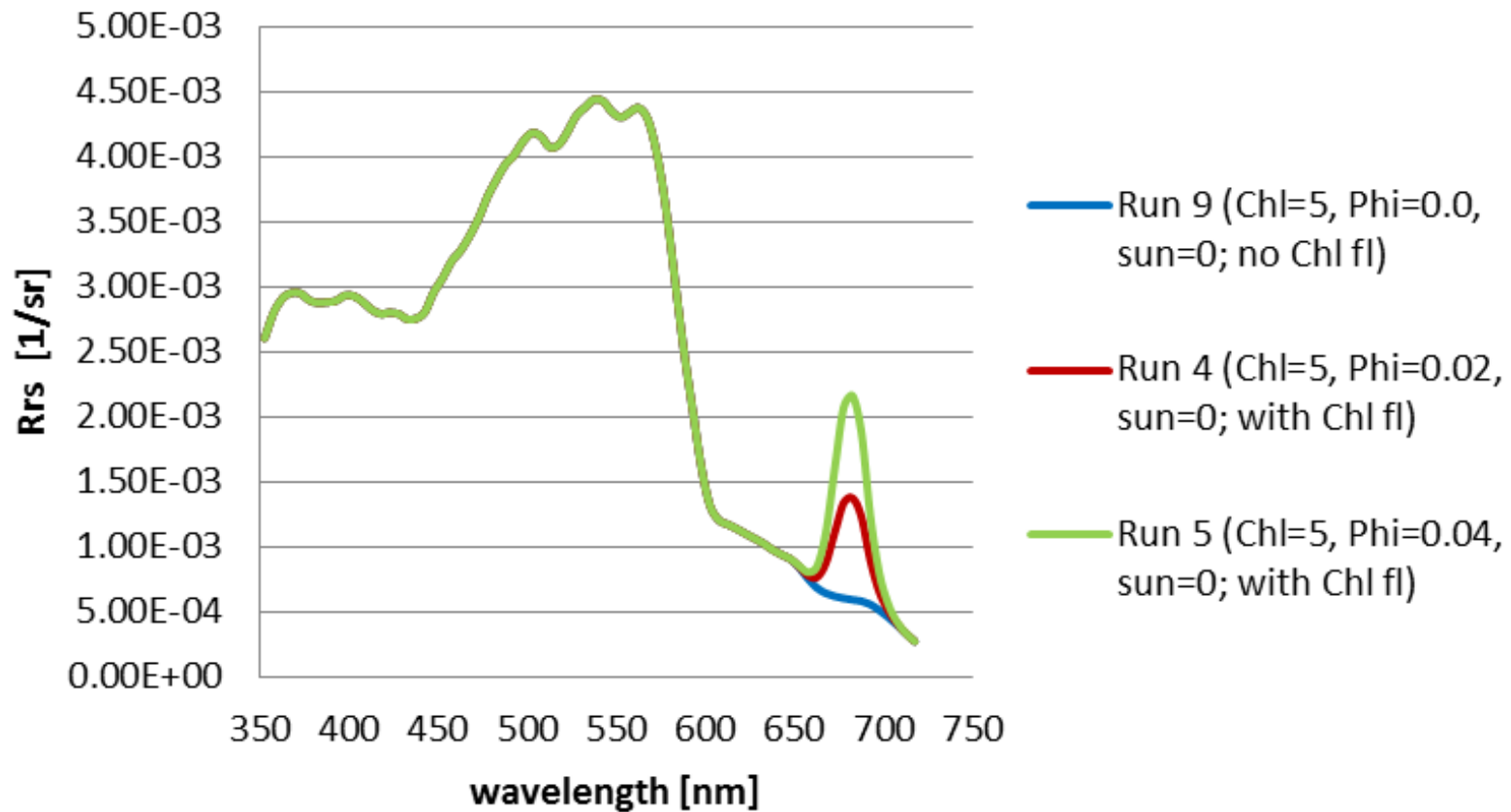


# What causes these differences?



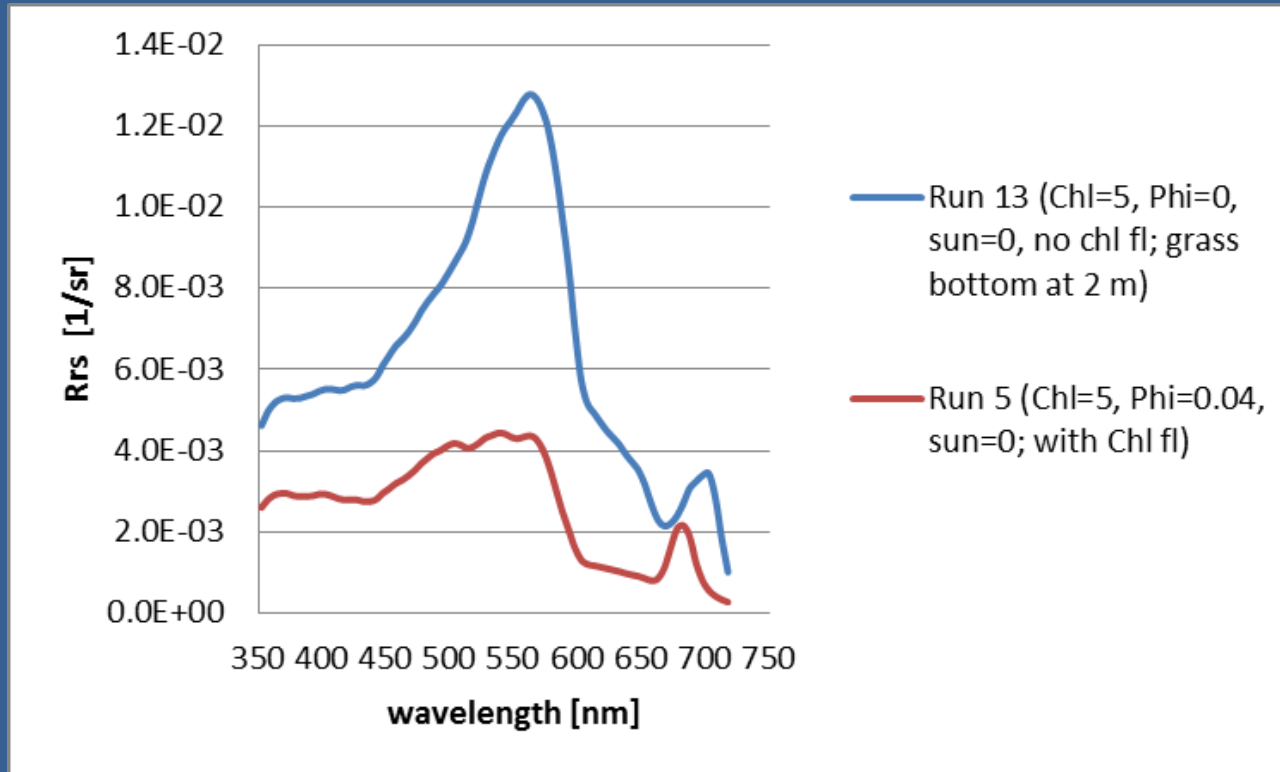
The sun angle effect is more noticeable for the high Chl case because there is more scattering and the phase function is different from the low Chl case

# What causes these differences?



Differences in Phi affect only the Chl fluorescence band

# Extra credit question: Explain the blue curve (red is Chl = 5, $\Phi_{\text{Chl}} = 0.04$ ; sun at 0)



The overall magnitude increase could be caused by increased scattering by non-absorbing particles, but that would not shift the peak of the Chl fl band. The max near 700 instead of 685 flags this as being something else. In this case, a grass bottom at 2 m depth

The rising “red edge” reflectance of the sea grass bottom at 2 m starts the rise in  $R_{rs}$ , which is then suppressed by the increasing water absorption

